Identification and Quantification of the N-Acetylcysteine Conjugate of Allyl Isothiocyanate in Human Urine after Ingestion of Mustard

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Abstract

Allyl isothiocyanate (AITC) is a constituent of cruciferous vegetables. It occurs widely in the human diet as a natural ingredient or food additive. AITC possesses numerous biochemical and physiological activities. It is cytotoxic and tumorigenic at high doses and also is a modulator of enzymes involved in metabolism of xenobiotics, including carcinogens. It is plausible that the wide consumption of dietary AITC may have profound effects on human health. To facilitate investigations of the effects of dietary AITC in humans, a method of measuring its uptake is needed. In this study, a urinary marker was developed for quantifying AITC uptake in humans. Four adult volunteers were asked to eat a meal containing brown mustard as the source of AITC. The 48-h urine samples were collected from these individuals and analyzed by reverse phase high performance liquid chromatography.

A major urinary metabolite was found, which was identified as N-acetyl-S-(N-allylthiocarbamoyl)-L-cysteine, the N-acetylcysteine conjugate of AITC, by comparing its retention time and UV, nuclear magnetic resonance, and mass spectra with those of the synthetic standard. After ingestion of mustard, the AITC conjugate was detected in urine collected from 0 to 12 h. No conjugate was found in urine samples collected after 12 h. The major portion of this metabolite was excreted within 8 h. The average total excretion of AITC conjugate was 5.4 ± 1.7 (SD) mg after consumption of 10 g of mustard and 12.8 ± 2.0 mg when 20 g of mustard was consumed. Thus, a dose-dependent excretion of this metabolite was demonstrated. The average conversion rate of AITC to its urinary N-acetylcysteine conjugate in humans was estimated to be 53.5 ± 8.1%. These results suggest that the urinary N-acetylcysteine conjugate of AITC may be a convenient and useful biomarker for quantifying human exposure to AITC.

Introduction

AITC is widely present in cruciferous vegetables such as cabbage, broccoli, kale, cauliflower, and horseradish. It is also commonly used in the human diet as a flavor agent. Like other isothiocyanates, AITC inhibits microsomal enzyme activities. Previous studies have shown that liver microsomes, obtained from rats that were fed a diet containing AITC, metabolize nitrosamines to a lesser extent than those of the untreated rats. AITC and its glucosinolate precursor, sinigrin, given in the diet, also inhibit hepatic DNA methylation induced by the tobacco-specific nitrosamine 4-(methylamino)-1-(3-pyridyl)-1-butanone in rats. These results suggest the potential of AITC in modulating the carcinogenic activities of nitrosamines, since many arylalkyl isothiocyanates structurally related to AITC are known to be inhibitors of lung tumorigenesis induced by 4-(methylamino)-1-(3-pyridyl)-1-butanone. It was shown recently that AITC inhibits the growth of human cancer cells in vitro. Furthermore, several authors have reported that AITC induces the Phase II detoxification enzyme glutathione S-transferases. On the other hand, chronic treatment with high doses of AITC induces urinary bladder tumors in rats. The diverse biochemical and biological activities of AITC and its wide consumption suggest its potential effects on human health.

Human exposure to AITC is mainly through the consumption of mustard, in particular brown mustard, and cruciferous vegetables. Because information on the exact content of AITC in these foods usually is not available and sometimes is impossible to obtain due to different storage and cultivation conditions, it is difficult to estimate the uptake of AITC in humans. Therefore, a marker would be useful to quantitatively monitor human exposure to AITC through diet. This information will be used to evaluate in epidemiological investigations the possible effects of dietary AITC on human health. Previously, it has been shown that the N-acetylcysteine conjugate of AITC (Fig. 1) is a urinary metabolite in rodents treated with AITC. The urinary metabolites of the structural analogues of AITC, BITC and PEITC, have been studied in humans. In this study, we describe the identification and use of the N-acetylcysteine conjugate of AITC as a simple and con-
convenient urinary marker for the uptake of AITC after a mustard meal.

**Materials and Methods**

**Instrumentation.** NMR spectra were recorded on a Bruker AM 360 WB spectrometer using methanol-d$_4$ as solvent. Negative ion desorption chemical ionization mass spectra were obtained on a Hewlett-Packard 5988A mass spectrometer. A HPLC system (Waters Associates, MA) equipped with an automatic gradient controller, two Model 501 pumps, and a Waters 990 photodiode array detector in conjunction with reverse-phase C$_{18}$ columns were used in the analyses and purification of the N-acetylcysteine conjugate of AITC. A Varian 3400 gas chromatograph equipped with a fused silica capillary column (60 m x 0.32 mm inside diameter, 1 µm thickness, DB-1; J & W, Inc.) and a flame ionization detector were used to analyze the concentration of AITC in the mustard paste.

**Chemicals.** AITC was purchased from Aldrich Chemical Co. (Milwaukee, WI) and N-acetylcysteine was purchased from Sigma Chemical Co. (St. Louis, MO). The N-acetylcysteine conjugate of AITC was prepared as described in the literature (18) and was characterized by 1H, 13C-NMR spectroscopy, and mass spectrometry. The measured chemical shifts (δ) and coupling constants (J) are given as: 1H-NMR (360 MHz, in CD$_3$OD, δ in ppm referenced to tetramethysilane) 2.00 (3H, s, CH$_3$), 3.55 (1H, dd, J: 14.1, 8.6 Hz, Cys-CH$_2$), 4.02 (1H, dd, J: 14.1, 4.7 Hz, Cys-CH$_2$), 4.28 - 4.40 (2H, ddd, J: 12.2, 5.6, 1.5 Hz, acetyl-CH$_2$), 4.72 (1H, dd, J: 8.6, 4.7 Hz, Cys-CH$_3$), 5.18 (1H, ddd, J: 10.2, 1.5, 1.5 Hz, cis-vinyl-CH$_2$), 5.26 (1H, ddd, J: 17.2, 1.5, 1.5 Hz, trans-vinyl-CH$_2$), 5.94 (1H, ddt, J: 17.2, 10.3, 5.7 Hz, vinyl-CH); 13C-NMR (92.52 MHz, in CD$_3$OD, δ in ppm referenced to tetramethysilane) 199.2 (C==S), 173.5, 173.2 (two C==O), 133.7 (CH=CH), 117.6 (CH$_3$==), 53.8 (Cys-CH$_2$), 50.4 (allyl-CH$_2$), 38.3 (Cys-CH$_3$), 22.8 (N-CH$_3$); MS (m/z) 58 (M+H); 261 (M+H); 244, 221, 162, 131 (base peak), 58. Grey Poupon Dijon mustard was purchased from a local grocery store and kept refrigerated after opening. Mustard was chosen as a source of AITC because it is frequently used in cooking and thus it is relatively convenient to use in human studies.

**Quantitative Analysis of AITC in Grey Poupon Country Dijon Mustard.** Grey Poupon Dijon Mustard paste (100 g), combined with tert-butyl isothiocyanate (14.36 mg) as an internal standard, was thoroughly mixed with 1000 ml of distilled water and 200 g of NaCl. The mixture was stirred with 200 ml of CH$_2$Cl$_2$ for 3 h and then filtered through Celite 545. After filtration, the CH$_2$Cl$_2$ phase was separated from the aqueous phase and subsequently dried over 20 g of anhydrous Na$_2$SO$_4$. After removing Na$_2$SO$_4$ by filtration, the CH$_2$Cl$_2$ extract was concentrated by a stream of N$_2$ gas. The concentrated extract was used in the gas chromatography analysis using the following conditions: injector temperature, 270°C; detector temperature, 300°C; helium carrier flow rate, 1 ml/min; temperature program, 40°C (5 min), 2°C/min, 260°C (20 min); split ratio, 50:1.

**Human Studies.** Two experiments using different amounts of mustard were performed. Each experiment involved four adult volunteers (two males and two females, age 20–45). In the first experiment, 10 g of mustard was ingested with bagel or bread at breakfast by each participant. All participants were advised to avoid cruciferous vegetables, mustard, and mustard flavored foods in the diet 2 days prior to and during the experiment. In the control experiment, all participants were asked to eat the same food as in the experimental diet with the only exception of mustard. In the second experiment, participant 1 in the first experiment was replaced by another volunteer of the same sex. The same protocol was used except that 20 g of mustard was consumed with turkey or chicken sandwiches in a lunch. In both experiments, urine samples were collected at intervals of 0–2, 2–4, 4–8, 8–12, 12–24, 24–36, and 36–48 h following breakfast or lunch. Urine samples were analyzed immediately or stored at −20°C overnight. After thawing, an aliquot (50 µl) of clear urine sample (the sample was centrifuged if not clear) was analyzed by a reverse phase HPLC system consisting of a Waters C$_{18}$-µBondapak column eluted isocratically with acetonitrile (10%) in 20 mM phosphate buffer (pH 3.0) at a flow rate of 1 ml/min.

**Quantification.** The HPLC peak of AITC conjugate detected at wavelength of 254 nm was used for integration. Standard solutions were prepared in 20 mM phosphate buffer (pH 3.0) with various concentrations of a synthetic N-acetylcysteine conjugate of AITC. The urinary metabolite was quantified with a calibration curve obtained using these standard solutions, which is linear over the concentration range examined (10$^{-6}$ to 10$^{-3}$ M). The urine samples were analyzed in the same fashion as the standards. Single and triple HPLC measurements were performed for samples obtained from experiments 1 and 2, respectively.

**Isolation and Identification of the N-Acetylcysteine Conjugate of AITC in Human Urine.** All crude urine samples collected in experiment 1 from 2–4 h following ingestion of mustard were combined (800 ml). Ammonium sulfate (160 g) was added and dissolved in the urine. The pH of the solution was adjusted to 3 with 12 N HCl. The acidic medium prevents possible decomposition of the conjugate during the work-up process. The solution was extracted with ethyl acetate (2 × 200 ml). The organic phase was washed twice with water (100 and 40 ml) and once with saturated NaCl solution (40 ml). After removing the solvent by a rotary evaporator under vacuum, the solid residue was dissolved in 5 ml of deionized water. Using a semipreparative reverse-phase C$_{18}$ HPLC column (Whatman Partisil 10 ODS-3 column Magnum 9), a mobile phase of 20% acetonitrile in 20 mM aqueous phosphate buffer (pH 3.0), and an isocratic elution at a flow rate of 2.5 ml/min, the AITC conjugate eluted at 30 min was purified and obtained in sufficient quantity after repetitive runs. The collected fractions were combined and evaporated to dryness under vac-
The N-acetylcysteine conjugate of AITC by comparing its to be N-acetyl-S-(N-allylthiocarbamoyl)-L-cysteine by corn-ods." It has the same characteristic UV absorptions as the conjugate. This compound was isolated and purified from coelutes with the synthetic standard of the N-acetylcysteine synthetic standard (Fig. 3). Its identity was further confirmed tracts were combined and the solvent was removed in a vacuum to afford a solid. The compound was identified as the AITC conjugate in the urine collected at different time intervals are shown in Table 1. Normally, the amount of mustard consumed, i.e., 5.4 ± 1.7 (SD) mg (3.6-7.6 mg) and 12.8 ± 2.0 mg (10.5-15.2 mg) are excreted corresponding to 10 and 20 g of mustard consumed. These results showed an uptake-dependent excretion of the AITC metabolite after mustard meals.

Consistent with the literature report (18), we have found that the N-acetylcysteine conjugate of AITC is in equilibrium with its free form. In our study, the equilibrium was evident by the presence of a small peak eluting after the conjugate which coeluted with AITC under the HPLC conditions used. A significant percentage of the N-acetylcysteine conjugate of AITC decomposed during an extended period of storage in a neutral medium, even at −20°C. Because of its instability, caution should be taken in quantifying the levels of this conjugate in the urine. Previously, we have found that PEITC is stabilized in acidic medium (16); it is likely that the AITC conjugate would be considerably more stable at acidic pH.

Although brown mustard is known to be rich in AITC (4), the exact content of AITC in the commercial products was not available. We have used gas chromatography to quantitatively analyze AITC in the mustard paste used in the human experiments. The result showed that the AITC content of the mustard is 453 ppm, or 0.453 mg of AITC/g of mustard. Using this information, we were able to calculate the conversion rate of AITC to its N-acetylcysteine conjugate in humans, as shown in Table 2. The average of the individual conversion rates is 53.5 ± 8.1%, which is consistent with the previous studies on the metabolism of BITC and PEITC in humans (16, 17). Those studies have shown that the conversion rates of these two isothiocyanates to their corresponding urinary N-acetylcysteine conjugates are 53.7 ± 5.9% and 47 ± 16%, respectively. Assuming that the average conversion rate of AITC obtained here is applicable to a larger population and is independent of the source of AITC, one may estimate the amount of AITC to which humans were recently exposed through the consumption of various foods and vegetables by simply measuring their urinary excretions of the N-acetylcysteine conjugate of AITC.

Discussion
The in vivo metabolism of several natural isothiocyanates has been studied in rodents and humans. For instance, the N-acetylcysteine conjugates of AITC and PEITC are excreted in the urine of mice (14, 19), although the major
metabolite of PEITC in mice is a cyclic mercaptopyruvic conjugate (19). However, the N-acetylcysteine conjugates are the major urinary metabolites in rats treated with AITC and BITC (15, 20). In humans, the N-acetylcysteine conjugates of BITC and PEITC appear to be the only urinary metabolites following ingestion of BITC, garden cress, and watercress (16, 17). However, to the best of our knowledge, the metabolism of AITC in humans has not been reported before.

Conjugations of isothiocyanates with glutathione appear to be the major metabolic pathway in humans, since most of their urinary metabolites are mercapturic acids or other derivatives from glutathione conjugates (14–19). Although the Phase II enzyme glutathione S-transferase-catalyzed conjugation of isothiocyanates is considered to be a natural detoxification process (16, 17), it has been postulated that this pathway may also be involved in the cyto-

**Table 1** Cumulative amounts of the N-acetylcysteine conjugate of AITC in human urine 12 h after ingestion of mustard

<table>
<thead>
<tr>
<th>Subject</th>
<th>Time interval (h)</th>
<th>Experiment 1</th>
<th>Experiment 2</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>AITC conjugate (mg)</td>
<td>Total excretion (mg)</td>
<td>AITC conjugate (mg)</td>
</tr>
<tr>
<td>1</td>
<td>0–2</td>
<td>0.7</td>
<td>2.8 ± 0.1</td>
</tr>
<tr>
<td></td>
<td>2–4</td>
<td>2.6</td>
<td>4.0 ± 0.2</td>
</tr>
<tr>
<td></td>
<td>4–8</td>
<td>2.4</td>
<td>3.2 ± 0.1</td>
</tr>
<tr>
<td></td>
<td>8–12</td>
<td>1.6</td>
<td>0.5 ± 0.1</td>
</tr>
<tr>
<td>2</td>
<td>0–2</td>
<td>2.6</td>
<td>4.7 ± 0.1</td>
</tr>
<tr>
<td></td>
<td>2–4</td>
<td>1.1</td>
<td>d</td>
</tr>
<tr>
<td></td>
<td>4–8</td>
<td>ND*</td>
<td>5.0 ± 0.1</td>
</tr>
<tr>
<td></td>
<td>8–12</td>
<td>ND*</td>
<td>3.6 ± 0.1</td>
</tr>
<tr>
<td>3</td>
<td>0–2</td>
<td>3.6</td>
<td>2.8 ± 0.1</td>
</tr>
<tr>
<td></td>
<td>2–4</td>
<td>d</td>
<td>6.1 ± 0.5</td>
</tr>
<tr>
<td></td>
<td>4–8</td>
<td>1.2</td>
<td>2.9 ± 0.2</td>
</tr>
<tr>
<td></td>
<td>8–12</td>
<td>0.4</td>
<td>0.5 ± 0.0</td>
</tr>
<tr>
<td>4</td>
<td>0–2</td>
<td>1.1</td>
<td>1.9 ± 0.2</td>
</tr>
<tr>
<td></td>
<td>2–4</td>
<td>1.2</td>
<td>5.4 ± 0.1</td>
</tr>
<tr>
<td></td>
<td>4–8</td>
<td>2.2</td>
<td>6.2 ± 0.6</td>
</tr>
<tr>
<td></td>
<td>8–12</td>
<td>0.4</td>
<td>1.7 ± 0.2</td>
</tr>
</tbody>
</table>

* Based on one measurement for each sample.

* Mean ± SD of three separate determinations.

* Subject 1 participated in experiment 1 but was replaced by another individual of the same sex in experiment 2.

* No urine was excreted during this period.

* ND, not determined due to peak overlap.

Fig. 4. (a) Comparison of the 360 MHz 1H-NMR of the synthetic standard of the N-acetylcysteine conjugate of AITC and the metabolite isolated from human urine samples after ingestion of mustard. Note that differences for solvent peak intensities at 3.35 and 4.97 ppm are due to the different concentrations of the two samples. The resonances for the AITC conjugate are identical in two spectra. (b) Comparison of their mass spectra. The major fragments C₆H₅NS₂ (131 m/z) and C₆H₅NO (58 m/z) for the AITC conjugate negative ion (261 m/z) were observed in both spectra.
Fig. 5. The concentration of the N-acetylcysteine conjugate of AITC in urine collected from four volunteers versus time (hours) following the ingestion of mustard in experiment 2.

**Table 2** Estimated percentage of conversion of allyl isothiocyanate to the N-acetylcysteine conjugate of AITC in human after a mustard meal

<table>
<thead>
<tr>
<th>Subject</th>
<th>Total conjugate excreted (mg)</th>
<th>AITC equivalent (mg)</th>
<th>Conversion (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>10.5</td>
<td>4.0</td>
<td>44.2</td>
</tr>
<tr>
<td>2</td>
<td>13.3</td>
<td>5.0</td>
<td>55.2</td>
</tr>
<tr>
<td>3</td>
<td>12.3</td>
<td>4.6</td>
<td>50.8</td>
</tr>
<tr>
<td>4</td>
<td>15.2</td>
<td>5.7</td>
<td>63.4</td>
</tr>
</tbody>
</table>

* Based on 0.453 mg/g weight of AITC in Grey Poupon mustard consumed in experiment 2.

The concentration of the N-acetylcysteine conjugate of AITC in urine collected from four volunteers versus time (hours) following the ingestion of mustard in experiment 2.

Toxicity of isothiocyanates (18). The glutathione conjugates of isothiocyanates are usually subject to further degradation to give final metabolites, the N-acetylcysteine conjugates of isothiocyanates, by enzymes such as γ-glutamyltranspeptidase, cysteinylglycinase, and N-acetyltransferase (20). Recently, the activities of the detoxification enzyme glutathione S-transferase have been associated with the risk of certain human cancers (21, 22). A survey of smokers demonstrated that individuals lacking glutathione S-transferase μ had a significantly higher incidence of lung cancer than those who display glutathione S-transferase μ activity (23). A discrepancy between phenotyping and genotyping the isozymes of glutathione S-transferase in relation to the risk of lung cancer in smokers was also reported (24, 25). The levels of excretion of the AITC conjugate in the urine following mustard consumption may be used to phenotype an individual for the activity of these enzymes. Therefore, it would be important to identify the specific glutathione transferase isozymes responsible for the conjugation of AITC.

It has been well documented that compounds in cruciferous vegetables induce Phase II detoxification enzymes, such as quinone reductase and glutathione S-transferase (11, 26). An isothiocyanate isolated from broccoli, (−)-1-isothiocyanato-(4R)-(methylsulfinyl)butane (CH$_3$SO-(CH$_2$)$_4$NCS, sulforaphane), was shown to be a strong Phase II enzyme inducer (27). Knowing that consumption of vegetables reduces the risk of cancer (28, 29), it is noteworthy that these isothiocyanates isolated from natural sources, including AITC, PEITC, and sulforaphane, may function as either Phase I enzyme inhibitors (5, 30, 31), which prevent the activation of carcinogens, and/or as Phase II enzyme inducers (11, 27). Much work is needed to further establish the detailed mechanism regarding how these naturally occurring compounds may work in humans to reduce the risk of cancer. The results presented here should provide a useful tool in the epidemiological investigations of the biological role of AITC in humans.

**Acknowledgments**

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**References**


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